

**FORMULATION AND EVALUATION STUDIES OF EMULGEL CONTAINING  
MEFENAMIC ACID**

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**ABSTRACT**

The objective of the present investigation was formulate & characterize the emulgel containing mefenamic acid. Mefenamic acid emulgel using HPMC and NaCMC as gelling agent were successfully developed. The developed formulations were then characterized for their physical appearance, pH, viscosity, spreadability, extrudability, drug content and *in vitro* release. The emulgel (CF1) prepared using HPMC as gelling agent, propylene glycol as co-surfactant and clove oil as penetration enhancer was found to show better result when compared to emulgel (CF2) prepared using NaCMC as gelling agent. In CF1, correlation coefficient of zero order kinetics was found to be  $R^2 = 0.995$  and that of first order kinetics was to be  $R^2 = 0.771$ . Higuchi plot was found to be linear with regression coefficient  $R^2 = 0.902$ . Hence it shows that the drug release follows zero order kinetics. To confirm exact mechanism of drug permeation from the emulgel, the data was fitted according to the Korsmeyer-Peppas model. The value of slope of the plot  $n$  gives an indication of the release mechanism. When  $n=1$ , the release is independent of time i.e., zero order, if  $n=0.5$  then the release is by Fickian diffusion. When  $n=0.5-1$ , diffusion is non-fickian and when  $n>1.0$  then it is super case II transport. The 'n' exponent value of optimized batch was found to be 0.651. Hence it shows non-fickian diffusion.

**KEYWORDS:** Mefenamic acid, emulgel, release kinetics.**INTRODUCTION**

New drug delivery systems development is largely based on promoting the therapeutic effect of a drug and minimizing its toxic effects by increasing the amount and persistence of drug in the vicinity of target cell and reducing the exposure of non target cells. Emulgels are emulsion, either of the oil-in-water or water-in-oil type, which are gelled by mixing with a gelling agent. Emulsified gel is the stable one and superior vehicle for hydrophobic or poorly water soluble drugs. Thus an emulgel is a combination of emulsion and gel. In recent years, there has been great interest in the use of novel polymers which can function as emulsifiers and thickeners because the gelling capacity of these compounds allows the formulations of stable emulsions and creams by decreasing surface and interfacial tension and the same time increasing the viscosity of aqueous phase. In fact, the presence of a gelling agent in water converts a classical emulsion into an emulgel.

Mefenamic acid is a non-steroidal anti-inflammatory drug which is widely used to treat pain, inflammation and in rheumatoid arthritis. Currently topical preparation of mefenamic acid is not available in the market. Emulgel is an emulsion either of oil-in-water or water-in-oil type which are gelled by mixing with a gelling agent; having favorable properties such as being removable

emollient, non-staining, water soluble, greater shelf life, bio-friendly, clear and pleasant appearance. The main aim of this project was to formulate an emulgel of mefenamic acid using different gelling agents like HPMC and NaCMC; thereby reducing side effects like ulceration, gastro intestinal disturbances, liver problems and kidney failure associated with its oral administration and to increase its patient acceptability.<sup>[1,5,8]</sup>

**MATERIALS AND METHODS**

Mefenamic acid pure drug was generously gifted by Camarin Pharmaceutical, Kannur. Sodium CMC & E was gifted by Degussa India Pvt. Ltd, Mumbai. Ethyl cellulose was purchased from Lobachemei Mumbai. All other excipients used in our work were of analytical grade.

**Determination of  $\lambda_{max}$** 

Dissolve accurately weighed 100mg of mefenamic acid in 100ml of methanol in 100ml standard flask to get 1000 $\mu$ g/ml. From the stock solution of mefenamic acid, 1ml is pipette out and diluted to 100ml with methanol to get 10 $\mu$ g/ml. The absorption maximum of the standard solutions of mefenamic acid was scanned between 200-400nm regions on UV-visible spectrophotometer. The absorption maxima obtained with the substance being

examined corresponds in position and relative intensity to those in the reference spectrum.

### Preformulation Studies<sup>[5]</sup>

Pre-formulation testing was an investigation of physical and chemical properties of a drug substance alone. It is the first step in rational development of dosage form.

### Solubility studies

Solubility of mefenamic acid was observed in different solvent such as distilled water in acetone, methanol, 95% ethanol, sodium hydroxide, Potassium hydroxide, diethyl ether, chloroform, acetic acid.

### Identification by melting point

Melting point of drug was determined using Melting point apparatus.

### Organoleptic properties

Physical appearance of drug was observed and compared with the official monographs.

### Partition Coefficient (Kp)

The partition coefficient of the drug was determined by shaking equal volumes of organic phase (n-octanol) and the aqueous phase in a separating funnel. A drug solution of 1 mg/ml was prepared in phosphate buffer pH7.4 and 50 ml of this solution was taken in a separating funnel and shaken with an equal volume of n-octanol for 10 minutes and allowed to stand for 24 hours with intermittent shaking. Then, the concentration was determined by U V Spectra.

**Drug-Excipient Interaction Studies<sup>6</sup>** in order to find out the possible interactions between mefenamic acid and the polymers used in the formulation of the Emulgel, Fourier transform infra-red spectroscopy (FT-IR) analysis was carried out on the pure substances and their physical mixtures.

**FT-IR Spectra** of the pure drug, NaCMC and HPMC and the physical mixture of the drug with polymers were taken individually by KBr pellet technique between 600 to 4000  $\text{cm}^{-1}$ . This is to ensure that there is no

incompatibility between the drug and the polymers. Once spectra were recorded, the peaks of the pure drug, the polymers and the physical mixture of drug and polymers were compared for any incompatibility.<sup>18,16,9</sup>

### Emulgel Formation

#### Effect of concentration of polymer

#### Gel Formation

The gels were prepared by dispersing gelling agent (HPMC or NaCMC) in different concentration in purified water with continuous stirring at moderate speed. Then the pH was adjusted to 6-6.5 using Triethanolamine

### Emulsion Formation

- The oil phase of emulsion were prepared by dissolving span 20 in light liquid paraffin while aqueous phase were prepared by dissolving Tween 20 in purified water.
- Methyl paraben and propyl paraben was added and mefenamic acid was mixed in ethanol and was mixed with aqueous phase.
- Oil phase and aqueous phase were then separately heated to 70-80°C.
- Oil phase were added to the aqueous phase with continuous stirring until cool to room temperature.

### Emulgel Formation

- The emulgel were obtained by mixing gel and emulsion in the ratio 1:1.

### Effect of co- surfactant

- To the selected formulations, different concentrations of co- surfactant was incorporated.

### Effect of penetration enhancer

- To the formulations showing best consistency, viscosity and release properties with presence of co-surfactant, 2 ml of penetration enhancer was added and further studies were done.<sup>[10,11]</sup>

The formulation component of microsponge emulgel is mentioned in table no: 1

**Table 1: Formulation of emulgel.**

Sl. No.	Ingredients	F1	F2	F3	F4	F5	F6
1	Mefenamic acid	0.1	0.1	0.1	0.1	0.1	0.1
2	HPMC	2.0	2.5	3.0	-	-	-
3	Sodium CMC	-	-	-	1.5	2.0	2.5
4	Light liquid paraffin	7.5	7.5	7.5	7.5	7.5	7.5
5	Span 20	0.5	0.5	0.5	0.5	0.5	0.5
6	Tween 20	1.3	1.3	1.3	1.3	1.3	1.3
7	Methyl paraben	0.001	0.001	0.001	0.001	0.001	0.001
8	Propyl paraben	0.003	0.003	0.003	0.003	0.003	0.003
9	Ethanol	5	5	5	5	5	5
10	Purified water	q.s	q.s	q.s	q.s	q.s	q.s

**Characterization of Mefenamic Acid Emulgel<sup>[8,15,21]</sup>****Physical Appearance**

The prepared emulgel formulations were inspected visually for their color, homogeneity, consistency, grittiness and phase separation.

**pH Determination**

1g of gel was accurately weighed and dispersed in 100 ml of distilled water. The pH of dispersion was measured by using digital pH meter

**Rheological studies**

Brookfield digital viscometer was used to measure the viscosity (in cps) of the prepared emulgel formulations. The spindle number 62 was rotated at 50 rpm for the viscosity measurement.

**Spreadability**

Spreadability of the formulation was determined by using an apparatus designed and developed in the laboratory especially for the project and diagram of the apparatus is shown in fig.5. Two rectangular glass plates of standard dimension were selected. 500mg of the sample was placed on one of the glass plate. Second plate was placed over the other one to sandwich sample between plates. A 20gm weight was placed on the top of upper plate to provide a uniform thin film of the sample between the plates. Weight was removed; excess of the gel sample was scrapped off from the edges. The top plate was then subjected to pull by using string to which 50gm weight was applied. The time required by the upper plate to travel a distance of 6cm and separate from the lower plate was noted. A shorter interval indicates better spreadability. Experiment was repeated and averages of three attempts were calculated for each formulation using following formula

$$\text{Spreadability} = \frac{(M \times L)}{T}$$

M = weight tied to the upper side

L = length of the glassslide

T = time in seconds.

**Extrudability**

The developed formulations were filled in collapsible metal tubes and crimped at one end. After removing the cap tube is pressed to extrude the product from the tube

**Drug content**

Drug content of the emulgel was determined by dissolving an accurately weighed quantity of 1g gel in about 100 ml of methanol. 2ml of this solution was diluted to 10ml with methanol. Solutions were then filtered and spectrophotometrically analyzed for drug content at 285nm. Drug content was determined from the standard curve of mefenamic acid.

**In vitro drug release of Mefenamic acid Emulgel****i) Activation of Egg membrane**

Activation of egg membrane was carried out by soaking the membrane sodium chloride saline solution to use. It was then mounted on the diffusion cell and equilibrated with receptor fluid for 15 minutes and used for drug release studies.

**ii) Drug Release studies<sup>[20]</sup>**

The *in vitro* release of mefenamic acid from the formulations were studied using modified Keshary-Chien apparatus which was fabricated in our laboratory and used for the release study. The dissolution medium used was Phosphate buffer 7.4 P<sup>H</sup>. 1 gm of the formulated emulgel was accurately weighed, and placed on membrane and placed in membrane and attached to this assembly. The donor compartment was suspended in 50 ml of dissolution medium maintained at 37± 1°C so that the membrane just touched the receptor medium surface. The medium was stirred at 50 rpm using magnetic stirrer. Aliquots, each of 1ml volume, were withdrawn at hourly intervals and replaced by an equal volume of the receptor medium. The aliquots were diluted to 10ml with the receptor medium and analyzed by UV-Visible spectrophotometer at 285 nm and % cumulative drug release was calculated.

**iii) Kinetics of drug release**

To examine the drug release kinetics and mechanism, the cumulative release data were fitted to models representing zero order, first, Higuchi's plot and Korsmeyer- Peppas plot respectively.

**RESULT**

**Determination of λ max:** Scanned in between 200-400 nm methanol as solvent maximum absorbance at 285 nm.

**Standard curve of mefenamic acid****Table 2: Standard curve of mefenamic acid.**

Concentration(µg/ml)	Absorbance
5	0.128±0.0015
10	0.253±0.0010
15	0.378±0.0021
20	0.496±0.0020
25	0.620±0.00208
30	0.735±0.00152

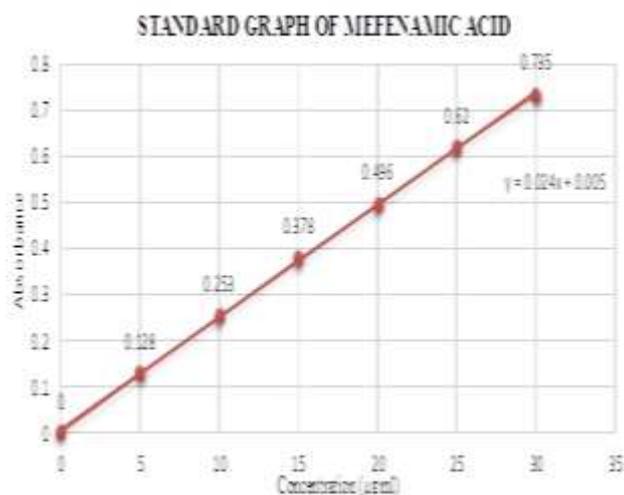


Figure no: 1 Calibration curve of mefenamic acid.

#### Identification & compatibility by FTIR

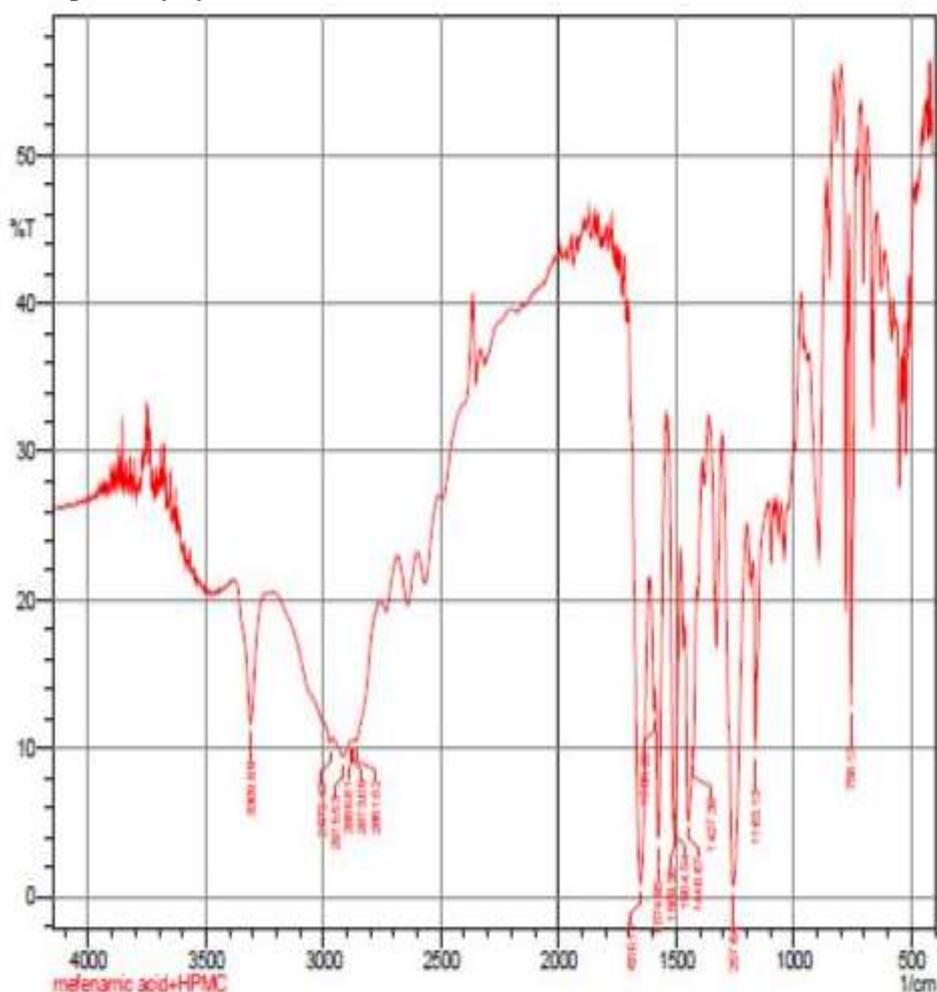


Figure 2: FTIR mefenamic acid-

#### Preformulation Studies

##### - Solubility profile

Solubility of drug has been carried out in different solvents and it is practically insoluble in water, slightly soluble in ethanol (95%) and soluble in alkali hydroxide.

##### - Determination of melting point

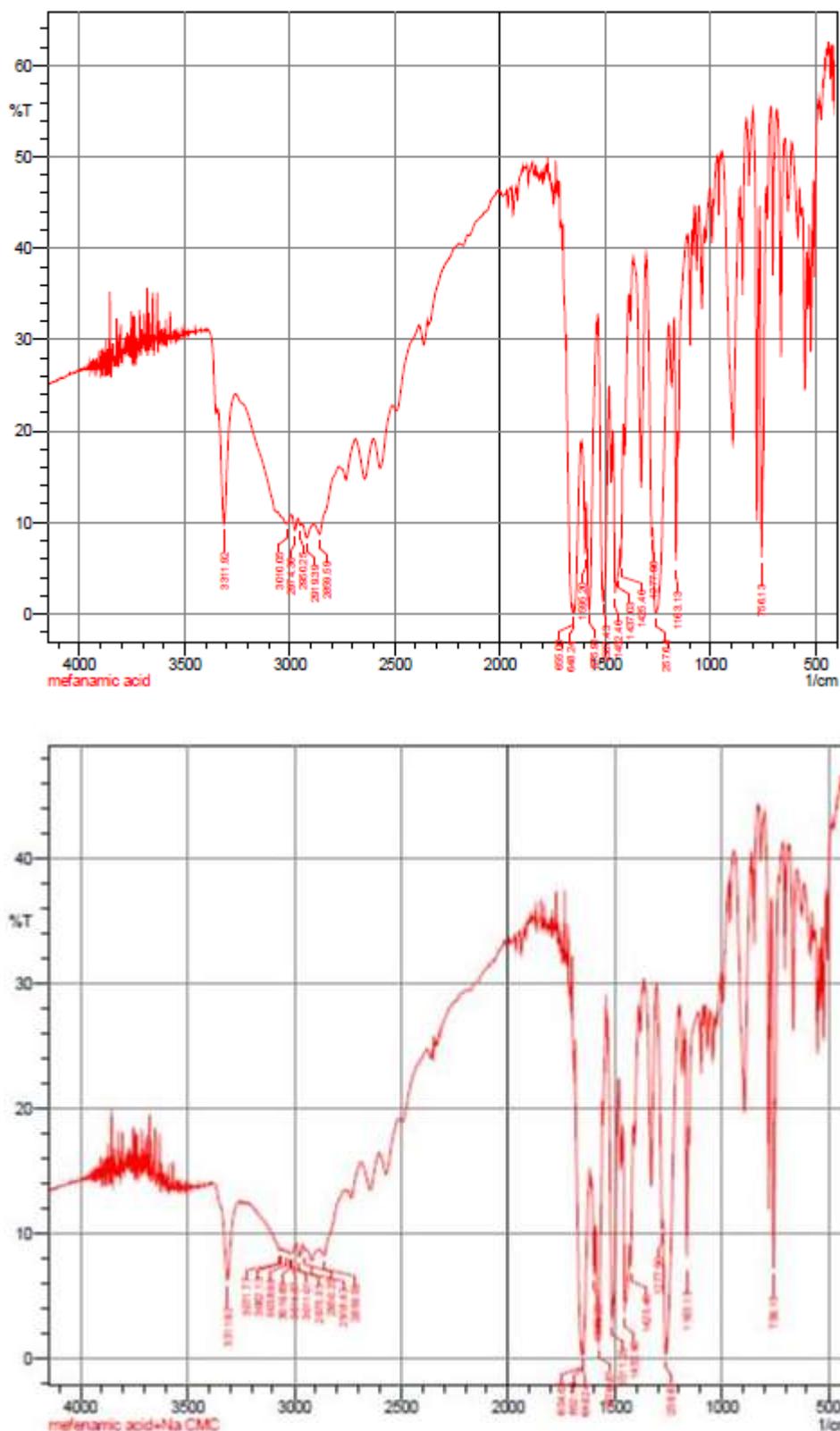
Melting point was determined using melting point apparatus. Temperature was noted at which solid drug changes in to a liquid and it was found to be 230-231°C.

##### Physical appearance

It occurs as white powder and it is odourless.

##### Partition coefficient

The partition coefficient was found to be 1.87



## Effect of Polymer Concentration On Emulgel Characterization of formulated emulgels

### 1. Physical evaluation

Table 3: Physical evaluation of formulations.

Formulation code	Color	Phase separation	Homogeneity	pH
F1	White	no	Excellent	6.0±1.25
F2	White	no	Excellent	6.4±1.47
F3	White	no	Excellent	6.5±0.98
F4	White	no	Excellent	6.4±1.14
F5	White	no	Excellent	6.7±1.37
F6	White	no	Excellent	6.5±1.47

### 2. Rheological Studies

Table 4: Rheological studies of formulations.

Formulation code	Spindle number	Revolutions per minute (RPM)	Torque (%)	Viscosity (cp)
F1	S63	50	88.2	1379±22.50
F2	S63	50	88.5	1425±31.1
F3	S63	50	82.8	1534±26.51
F4	S63	50	81.5	1474±33.4
F5	S63	50	86.3	1581±31.39
F6	S63	50	88.1	1612±33.4

### 3. Extrudability

Table 5: Extrudability of formulations.

Formulation code	Weight extruded from the tube (gm)
F1	0.69±.128
F2	1.02±0.20
F3	1.1±0.096
F4	0.73±0.121
F5	0.81±0.15
F6	0.89±0.151

### Spreadability

Table 6: Spreadability coefficient of formulations.

Formulation code	M(gm)	L(cm)	T(sec)	Spreadability
F1	50	6	10	30±0.54
F2	50	6	11	23.07±0.47
F3	50	6	15	20±0.34
F4	50	6	10	30±0.46
F5	50	6	11	27.27±0.65
F6	50	6	16	21.42±0.24

### Drug content

Drug content of the formulated emulgels were estimated by UV spectrophotometer at  $\lambda_{max}$  285 nm and the results of drug content of each formulation was given in the table below.

Table 7: Drug content of emulgels.

Product	Drug content (%)
F1	84.21±1.35
F2	85.21±1.16
F3	84.29±0.98
F4	73.45±0.87
F5	75.88±1.31
F6	74.88±1.10

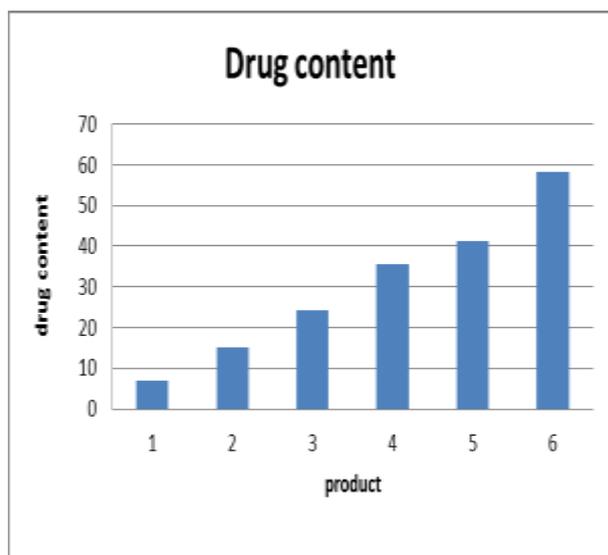
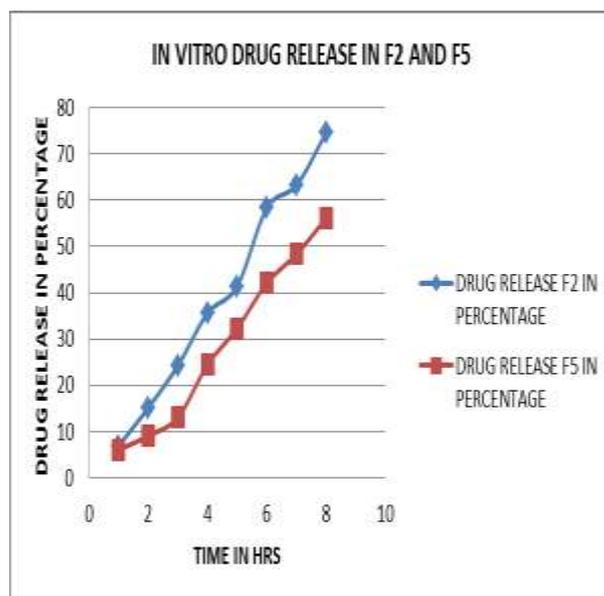


Figure 4: Drug content of emulgel.



**In vitro Drug release study**

From the above studies it was seen that consistency and viscosity of formulations F2& F5 were the best and the *in vitro* studies of above formulations were conducted.

**Effect of Co-Surfactant Concentration**

To the selected formulations of emulgels different concentration of co- surfactant were added and further studied.

Table 8: Cumulative percentage drug release of F2 & F5.

Time (hrs.)	Cumulative percentage drug release	
	F2	F5
1	7.0	6.1
2	15.2	9.1
3	24.3	13.1
4	35.6	24.1
5	41.4	32.3
6	58.3	42.1
7	63.2	48.3
8	74.5	56.1

Table 9: Formulation of emulgels containing co-surfactants.

Sl. No	Ingredients	F7	F8	F9	F10
1	Mefenamic acid	0.1	0.1	0.1	0.1
2	HPMC	2.5	2.5	-	-
3	Sodium CMC	-	-	2.0	2.0
4	Light liquid paraffin	7.5	7.5	7.5	7.5
5	Span 20	0.5	0.5	0.5	0.5
6	Tween 20	1.3	1.3	1.3	1.3
7	Propylene glycol	2.5	5.0	2.5	5.0
7	Methyl paraben	0.001	0.001	0.001	0.001
8	Propyl paraben	0.003	0.003	0.003	0.003
9	Ethanol	5	5	5	5
10	Purified water	q.s	q.s	q.s	q.s

**Characterizations of emulgels**

Table 10: evaluation of formulations (F7, F8, F9& F10).

Formulation code	Color	Phase separation	homogeneity	pH	Viscosity	Weight extruded from the tube (gm)
F7	white	no	Excellent	6.4±0.98	1420± 21.10	1.15±0.20
F8	white	no	Excellent	6.5±0.78	1415± 31.10	1.03±0.79
F9	white	no	Excellent	6.5±0.65	1538 ±27.5	0.93±0.09

F10	white	no	Excellent	6.8±0.71	1520± 21.10	1.2±0.114
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**Spreadability****Table 11: Spreadability of formulations (F7, F8, F9& F10).**

Formulation code	M(gm)	L(cm)	T(sec)	MLT
F7	50	6	13	23.07±1.82
F8	50	6	12	25.0±1.43
F9	50	6	14	21.42±1.28
F10	50	6	13	23.07±1.12

**5. Drug content****Table 12: Drug content of formulations (F7, F8, F9& F10).**

Formulation code	Drug content (%)
F7	86.71±2.86
F8	87.90±3.13
F9	76.45±3.46
F10	76.72±2.35

**6. *In vitro* Drug release study****Table 13: Cumulative percentage drug release of F7, F8, F9, & F10.**

Time (hrs)	Cumulative percentage drug release			
	F7	F8	F9	F10
0	0	0	0	0
1	7.9	8.4	6.1	6.34
2	15.5	16.8	9.2	10.7
3	24.6	25.1	16.1	18.93
4	38.4	40.5	29.6	33.2
5	43.1	47.3	32.3	41.6
6	58.5	60.7	44.5	51.5
7	67.5	68.21	54.1	58.4
8	76.5	79.6	61.3	63.4

**Effect of Penetration Enhancers**

From the above studies, formulations F8 and F10 were found to show better result, so to this formulations 2ml

of penetration enhancer (clove oil) were incorporated and evaluations were done.

**Characterizations of emulgels****Table 14: Evaluation of CF1& CF.**

Formulation code	Color	Phase seperation	homogeneity	pH	Viscosity	Weight extruded from the tube (gm)
CF1	Whitish Yellow	no	Excellent	6.5±0.98	1414±31.10	1.1±0.096
CF2	Whitish Yellow	no	Excellent	6.4±0.78	1519±21.10	1.02±0.20

**Spreadability****Table 15: Spreadability of CF1& CF.**

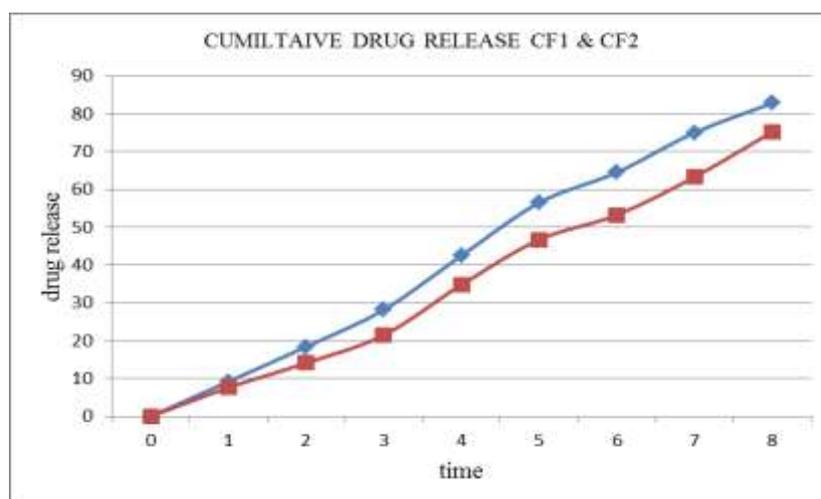
Formulation code	M(gm)	L(cm)	T(sec)	Spreadability
CF1	50	6	12	25.0±1.54
CF2	50	6	13	23.07±1.31

**Drug content****Table 16: Drug content of CF1 and CF2.**

Formulation code	Drug content (%)
CF1	87.90±3.14
CF2	76.72±4.78

**In vitro Drug release study****Table 17: Cumulative percentage drug release of CF1 & CF2.**

Time (hrs)	Cumulative percentage drug release	
	CF1	CF2
0	0	0
1	9.2	7.6
2	18.4	14.1
3	28.2	21.5
4	42.5	34.8
5	56.5	46.7
6	64.5	53.25
7	75.0	63.3
8	82.8	75.2



**Kinetics of drug release** was studied to examine the drug release kinetics and mechanism, the cumulative release data were fitted to models representing

- Zero order

- First order
- Higuchi's plot
- Korsmeyer Peppas model

**Drug release kinetics****Table 18: Drug release kinetics.**

Formulation code	Zero order (R2)	First order (R2)	Higuchi (R2)	Korsmeyer – peppas (R2)
CF1	0.995	0.771	0.902	0.651

**DISCUSSION**

The goal of any drug delivery system is to provide a therapeutic amount of drug to the proper site in the body and then maintain the desired drug concentration. A well designed drug delivery system can overcome some of problems of conventional therapy and enhance therapeutic efficacy of the given drug. There were various approaches in delivering therapeutic substance to the target site in sustained and controlled release fashion. One such approach is emulgel. In this study mefenamic acid emulgels was formulated using two polymers HPMC and NaCMC and their evaluations are performed and the results obtained.

Drug identification was done by performing melting point determination and FT-IR studies. From the result the melting point of drug was found to be 2300C which complies with official standard indicating the purity of

the sample. FT-IR studies peak of mefenamic acid obtained at  $3311.92\text{cm}^{-1}$ ,  $2974.36\text{cm}^{-1}$ ,  $1648\text{cm}^{-1}$ ,  $1595.2\text{cm}^{-1}$ ,  $1575.91\text{cm}^{-1}$ ,  $1507\text{cm}^{-1}$ ,  $1257\text{cm}^{-1}$ ,  $1163\text{cm}^{-1}$ ,  $756.13\text{cm}^{-1}$  showed that the peaks are identical to reference indicating the identity of drug. The FT-IR spectrums of pure drug, polymers and physical mixture of drug and polymers. (Figure) and (table) shows that no interaction took place between drug and polymer. However, some additional peaks were observed with physical mixtures, which could be due to the presence of polymers. These results suggest that there is no interaction between the drug and polymers used in the study. Thus indicating that the drug and polymer are compatible with each other

## Formulation and Characterizations of mefenamic acid emulgel

### Effect of polymer concentration

Three emulgel containing HPMC and three emulgel containing NaCMC as polymer were formulated.

### Characterizations of emulgels

#### 1. Physical appearance

All the prepared emulgel formulations were white viscous creamy preparations with a smooth and homogeneous appearance.

#### 2. pH determination

The pH values of all prepared formulations ranged 6.0-6.7, which lies within the normal pH range of the skin and was considered acceptable to avoid any irritation upon application to the skin.

#### 3. Rheological studies

The rheological behavior of all formulated emulgels was studied using Brookfield viscometer at a speed of 50rpm and spindle no.62 were used. The viscosity of the formulation increases as concentration of polymer increases.

#### 4. Extrudability

The extrusion of the emulgel from the tube is important during its application. Gel with high consistency may not extrude from the tube where as low viscous gel may flow quickly and hence suitable consistency is required to extrude the gel from the tube. Extrudability of HPMC emulgels were found to be good.

#### 5. Spreadability

The value of spreadability indicates that emulgels is spreadable by small amount. Spreadability of HPMC emulgels was in the range of 20-30gm.cm/sec and spreadability of NaCMC was in the range 21-30gm.cm/sec, indicating the spreadability of both type emulgel formulations were good. It was found that the spreadability increased with decreased viscosity.

#### 6. Drug content

Drug content of the formulated gels was estimated by UV spectrophotometer at  $\lambda_{max}$  285nm and drug content was calculated from calibration curve. Drug content of the formulations showed that the drug was uniformly distributed in to emulgels.

#### 7. *In vitro* Drug release studies

From the prepared six formulations, the formulations (F2&F5) showing good viscosity, extrudability, spreadability and drug content uniformity were selected. These formulations were then subjected to in-vitro release studies.

- For F2, in the first hour about 7.0% and at the end of 8<sup>th</sup> hour about 74.5% cumulative amount of drug was released.

- For F5, in the first hour about 6.1% and at the end of 8<sup>th</sup> hour about 56.1% cumulative amount of drug was released.

### Effect of co- surfactant concentration

From the above studies it was seen that the consistency and viscosity of formulations F2 & F7 were the best. Hence different concentration of co-surfactant ( propylene glycol) were added to two formulation and further studied.

### Characterizations of emulgels

#### 1. Physical evaluations

All the prepared emulgel formulations were white viscous creamy preparations with a smooth and homogeneous appearance. The pH values of all prepared formulations ranged 6.0-6.7, which lies within the normal pH range of the skin and was considered acceptable to avoid any irritation upon application to the skin.

#### 2. Rheological studies

The rheological behavior of all formulated emulgels was studied using Brookfield viscometer at a speed of 50rpm and spindle no.62 was used. The viscosities of the formulations were satisfactory.

#### 3. Extrudability

Extrudability of emulgels containing HPMC as polymer were found to be better than that of emulgels containing NaCMC

#### 4. Spreadability

The value of spreadability indicates that emulgels is spreadable by small amount. Spreadability of HPMC emulgels was in the range of 23-25gm.cm/sec and spreadability of NaCMC was in the range 21-24gm.cm/sec, indicating the spreadability of both type emulgel formulations were good.

#### 5. Drug content

Drug content of the formulated gels was estimated by UV spectrophotometer at  $\lambda_{max}$  285nm and drug content was calculated from calibration curve. Drug content of the formulations showed that the drug was uniformly distributed in to emulgels.

#### 6. *In vitro* Drug release studies

All the above formulations (F7, F8, F9 and F10) were then subjected to in-vitro release studies.

1. For F7, in the first hour about 7.9% and at the end of 8<sup>th</sup> hour about 76.5% cumulative amount of drug was released.
2. For F8, in the first hour about 8.4% and at the end of 8<sup>th</sup> hour about 79.6% cumulative amount of drug was released.
3. For F9, in the first hour about 6.1% and at the end of 8<sup>th</sup> hour about 61.3% cumulative amount of drug was released.

4. For F10, in the first hour about 6.34% and at the end of 8<sup>th</sup> hour about 63.4% cumulative amount of drug was released.

#### Above results indicates that

- As the polymer concentration increases emulgel become viscous but release decreases. So an optimum concentration of polymer which gives correct consistency and maximum drug release should be selected.
- As the concentration of co-surfactant increases the drug release also increases.

#### Effect of penetration enhancer

To the best formulations (F8 and F10) from above results, clove oil as penetration enhancer was added and effect was studied.

#### 1. Physical evaluations

The prepared emulgel formulations were whitish yellow creamy preparations with a smooth and homogeneous appearance. The pH values of CF1 and CF2 were 6.4 and 6.5 respectively, which lies within the normal pH range of the skin and was considered acceptable to avoid any irritation upon application to the skin.

#### 2. Rheological studies

The rheological behavior of all formulated emulgels was studied using Brookfield viscometer at a speed of 50rpm and spindle no.62 was used. The viscosities of CF1 and CF2 were 1379±22.50cp and 1425±31.1cp respectively, which shows that prepared emulgels have required consistency.

#### 3. Spreadability

The spreadability coefficient of CF1 and CF2 were 25.0 gm.cm/sec and 23.07 gm.cm/sec respectively. This result indicates CF1 had good spreadability than CF2.

#### 4. Extrudability

Both the emulgels had good extruding property. Comparatively CF1 has good extruding property than CF2.

#### 5. Drug content

Drug content of the formulated emulgels was estimated by UV spectrophotometer at  $\lambda_{max}$  285nm and drug content was calculated from calibration curve. Drug content of the formulations showed that the drug was uniformly distributed in to emulgels.

#### 6. *In vitro* release studies

- For CF1, in the first hour about 9.2% and at the end of 8<sup>th</sup> hour about 82.8% cumulative amount of drug was released.
- For CF2, in the first hour about 7.6 % and at the end of 8<sup>th</sup> hour about 75.2% cumulative amount of drug was released.

From above studies it was found that CF1 was an excellent emulgel and hence the data obtained from *in-vitro* release studies was fitted into various kinetic models and also stability studies were conducted on the selected formulation as per ICH guidelines.

#### Drug Release kinetics

In CF1, correlation coefficient of zero order kinetics was found to be  $R^2 = 0.995$  and that of first order kinetics was found to be  $R^2=0.771$ . Higuchi plot was found to be linear with regression coefficient  $R^2=0.902$  Hence it shows that the drug release follows zero order kinetics. To confirm exact mechanism of drug permeation from the emulgel, the data was fitted according to the Korsmeyer-Peppas model. The value of slope of the plot  $n$  gives an indication of the release mechanism. When  $n=1$ , the release is independent of time i.e., zero order, if  $n=0.5$  then the release is by Fickian diffusion. When  $n= 0.5-1$ , diffusion is non-fickian and when  $n>1.0$  then it is super case II transport. The 'n' exponent value of optimized batch was found to be 0.651. Hence it shows non-fickian diffusion.

#### CONCLUSION

Mefenamic acid emulgel using HPMC and NaCMC as gelling agent were successfully developed. The developed formulations were then characterized for their physical appearance, pH, viscosity, spreadability, extrudability, drug content and *in vitro* release. The emulgel (CF1) prepared using HPMC as gelling agent, propylene glycol as co-surfactant and clove oil as penetration enhancer was found to show better result when compared to emulgel (CF2) prepared using NaCMC as gelling agent.

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